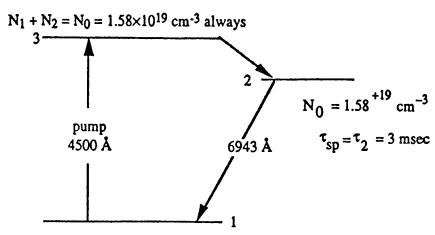
One should always start each problem with an energy level diagram showing how the atoms get to the upper state and how the atoms return. The most important fact is that atoms must be conserved and thus:



For $N_2 = 10^{19}$ cm⁻³, $N_1 = 0.58 \times 10^{+19}$ cm⁻³ assuming that nothing resides in state 3.

(b)
$$P_{spont} = \frac{N_2 V}{\tau_{sp}} \cdot hv_{laser} = 14.3 \text{ kW};$$

This power must be supplied by absorption on the $1 \rightarrow 3$ route. In the best of all worlds, all of the atoms pumped to 3 would transfer to state 2 – the fraction that does so is called the quantum efficiency of the pump. For ruby, that efficiency is 70%, but for this problem, we can assume it to be 100%

(a)
$$P_{pump} = \frac{\lambda_{31}}{\lambda_{21}} P_{spont} = 22.1 \text{ kW.}$$
 (It would be 31.6 kW for the real system)
 $\tau_{R.T.} = \frac{2l_g n_g}{c} + \frac{2(d - l_g)}{c} = \frac{2 \times 15 \times 1.78}{c} + \frac{2 \times 5}{c} = 2.1 \times 10^{-9} \text{ s};$
 $2g_{th} = \ln \frac{1}{R_1} + \ln \frac{1}{R_2} = 5.13 \times 10^{-2} + 0.357 = 0.408$

The first log term is internal loss (per round-trip), last is the external value. The photon lifetime is given by:

$$\frac{1}{\tau_p} = \frac{<\alpha_T>2d}{\tau_{R.T.}} = 19.4^{+9} \text{ s}^{-1}; \ \tau_p = 5.15 \text{ ns}; \ \eta_{cpl} = 0.357/0.408 = 0.875$$

$$P(\max) = \eta_{cpl} \cdot \frac{hv}{\tau_p} \left\{ \frac{n_i - n_{th}}{2} - \frac{n_{th}}{2} \ln \left(\frac{n_i}{n_{th}} \right) \right\}$$

 $n_i = 15 \text{ cm} \times 1 \text{ cm}^2 \text{ x} [(10^{19} - 0.58 \times 10^{19}) \text{cm}^{-3}] = 6.3 \times 10^{+19} \text{ atoms}$

$$\gamma_{th} = \frac{1}{2l_g} \ln \frac{1}{R_1 R_2} = 1.36^{-2} \text{ cm}^{-1} = (N_2 - N_1)_{th} \sigma;$$

 $(N_2 - N_1)_{th} \cdot A \cdot I_g = n_{th} = 1.61^{+19}$ atoms; Thus $n_1/n_{th} = 3.92$; and $\eta_{xtn} = 0.98$;

(c) Hence P(max) =
$$\frac{0.351}{0.408} \cdot \frac{2.86 \times 10^{-19}}{5.15 \times 10^{-9}} \{ 1.24 \times 10^{+19} \} = 603 \text{ MW}$$

(d)
$$W_{out} = \eta_{cpl} \eta_{xtn} \cdot (h v_n / 2) = 7.72$$
 Joules